Vibrational Spectroscopy

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Two similar structures Zincblende and diamond dispersion LO/TO splitting DFPT with LO/TO splitting in NaCl LO/TO splitting in non-cubic systems The non-analytic term and Born Charges Electric Field response in CASTEP Ionic and Electronic Dielectric Permittivity Example: NaHF2 Dynamical charges in NaHF2

Modelling of spectra

LO/TO Splitting



Two similar structures

LO/TO Splitting

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Modelling of spectra



Zincblende BN



Diamond



Zincblende and diamond dispersion





Cubic symmetry of Hamiltonian predicts *triply degenerate* optic mode at Γ in both cases.

2+1 optic mode structure of BN violates group theoretical prediction. Phenomenon known as **LO/TO splitting**.



LO/TO Splitting

Two similar structures Zincblende and diamond dispersion

LO/TO splitting

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Modelling of spectra



- Dipole created by displacement of charges of long-wavelength LO mode creates induced electric field.
- For TO motion $\boldsymbol{E} \perp \boldsymbol{q} \Rightarrow \boldsymbol{E}.\boldsymbol{q} = 0$
- For LO mode $E.q \neq 0$ and E-field adds additional restoring force.
- Frequency of LO mode is upshifted.
- Lyndane-Sachs-Teller relation for cubic case: $\frac{\omega_{LO}^2}{\omega_{TO}^2} = \frac{\epsilon_0}{\epsilon_{\infty}}$
- LO frequencies at q = 0 depend on *dielectric permittivity*



DFPT with LO/TO splitting in NaCl

LO/TO Splitting

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LO/TO splitting

DFPT with LO/TO splitting in NaCl

LO/TO splitting in non-cubic systems The non-analytic term and Born Charges Electric Field response in CASTEP Ionic and Electronic Dielectric Permittivity Example: NaHF2 Dynamical charges in NaHF2

Modelling of spectra

LO modes can be seen in infrared, INS, IXS experiments, but not raman. NaCl phonon dispersion





LO/TO splitting in non-cubic systems





The non-analytic term and Born Charges

LO/TO Splitting

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Electric Field response in CASTEP Ionic and Electronic Dielectric Permittivity Example: NaHF2 Dynamical charges in NaHF2

Modelling of spectra

LO-TO splitting at $q \rightarrow 0$ is automatically included in DFPT. At q = 0 exactly, need to add additional *non-analytic* term to dynamical matrix

$$C_{\alpha,\alpha'}^{\kappa,\kappa'}(\boldsymbol{q}=0)(NA) = \frac{4\pi}{\Omega_0} \frac{\sum_{\gamma} q_{\gamma} Z_{\kappa,\gamma\alpha}^* \sum_{\gamma'} q_{\gamma'} Z_{\kappa',\gamma'\alpha'}^*}{\sum_{\gamma\gamma'} q_{\gamma} \epsilon_{\gamma\gamma'}^{\infty} q_{\gamma'}}$$

 $\epsilon^{\infty}_{\gamma\gamma'}$ is the dielectric permittivity tensor $Z^*_{\kappa,\beta\alpha}$ is the Born Effective Charge tensor

$$Z^*_{\kappa,\beta,alpha} = V \frac{\partial P_{\beta}}{\partial x_{\kappa,\alpha}} = \frac{\partial F_{\kappa,\alpha}}{\partial E_{\beta}}$$

 Z^* is *polarization per unit cell* caused by displacement of atom κ in direction α or *force exerted on ion* by macroscopic electric field.

 $Z^*_{\kappa,\beta\alpha}$ and $\epsilon^{\infty}_{\gamma\gamma'}$ can both be computed via DFPT response to *electric field* perturbation.

(Dipole-dipole model use for Coulombic tail correction in Fourier interpolation procedure also depends on Z^* and ϵ^{∞} . Not included in most supercell calculations \Rightarrow be suspicious of convergence in case of polar systems).



LO/TO splitting

splitting in NaCl LO/TO splitting in

non-cubic systems

The non-analytic term and Born Charges

Ionic and Electronic Dielectric Permittivity Example: NaHF2 Dynamical charges in

Modelling of spectra

Electric Field response in

DFPT with LO/TO

dispersion

CASTEP

NaHF2

Two similar structures

Zincblende and diamond

- Can also apply DFPT to *electric field* perturbation.
 Need trick to evaluate position operator. Evaluate ∇_kφ. See NMR lecture for
 - details.
 - set task = efield or task = phonon+efield
 - Convergence controlled by efield_energy_tol
 - Computes dielectric permittivity for crystals and polarisability (for molecules)
 - Includes *lattice contribution* for $\omega \to 0$ response.
 - Writes additional file seedname.efield containing $\epsilon_{\gamma\gamma'}(\omega)$ in infrared region.

Optical Per	rmittivity (:	f->infinity)	DC	(f=0)	
3.52475	0.00000	0.00000	10.00569	0.00000	0.00000
0.00000	3.52475	0.00000	0.00000	10.00569	0.00000
0.00000	0.00000	3.52475	0.00000	0.00000	10.00569

		Pola	risabilities (A*	*3)	
	Optical (f	->infinity)		Static	(f=0)
5.327	0.000	0.00000	19.00151	0.00000	0.00000
0.000	5.327	0.00000	0.0000	19.00151	0.00000
0.000	0.000	5.32709	0.00000	0.00000	19.00151



Ionic and Electronic Dielectric Permittivity

LO/TO Splitting

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Example: NaHF2 Dynamical charges in

NaHF2

Modelling of spectra

Pure covalent system	m - Si	$Z^{*} = -$	-0.009,				
Ionic system, NaCl. α -quartz, SiO ₂ :	$\epsilon_0 = 6$	$5.95, \epsilon_{\odot}$	$_{\circ} = 2.70.$	$Z^*(Na) =$	1.060,	$Z^*(CI) = -1.05$	8
	0.44	0.00	0.00	5 01	0.00	0.00	

	2.44	0.00	0.00		5.01	0.00	0.00
$\epsilon_{\infty} =$	0.00	2.44	0.00	$\epsilon_0 =$	0.00	4.71	0.15
	0.00	0.00	2.49		0.00	0.15	4.58

(exp: $\epsilon_0 = 4.64/4.43$)

	-2.16	-0.03	0.81		2.98	0.00	0.00
$Z^*(O_1) =$	-0.08	-1.04	0.14	$Z^*(Si) =$	0.00	3.67	0.26
	0.00	0.00	-1.71		0.00	-0.30	3.45

Note anisotropic tensor character of Z^* .



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Dielectric Permittivity Example: NaHF2

Dynamical charges in NaHF2

Modelling of spectra

- Unusual bonding, nominally ionic Na⁺ FHF⁻, with linear anion.
- Layer structure, $R\bar{3}$ space group.
- Phase transition to monoclinic form at \approx 0.4 GPa.
- Phonon spectrum measured at ISIS on TOSCA
 - high-resolution neutron powder spectrometer.
 - No selection rule absences
 - Little or no anharmonic overtone contamination of spectra
 - No control over (q, ω) path excellent for molecular systems but spectra hard to interpret if dispersion present.





Dynamical charges in NaHF2

LO/TO Splitting

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Dynamical charges in NaHF2

Modelling of spectra

• highly anisotropic Z^*

xy	z	
0.34	2.02	(H)
-0.69	-1.52	(F)
1.06	1.04	(Na)

- charge density n⁽¹⁾(r) from DFPT shows electronic response under perturbation
- Charge on F ions moves in response to H displacement in z direction.





Modelling of spectra
Inelastic Neutron
Scattering
INS of Ammonium
Fluoride
Single-crystal infrared
Powder infrared
Modelling of powder IR
Non-resonant raman
Raman scattering of
t-ZrO2
Inelastic X-Ray
scattering
IXS of Diaspore

Modelling of spectra



Modelling of spectra Inelastic Neutron Scattering INS of Ammonium Fluoride Single-crystal infrared Powder infrared Modelling of powder IR Non-resonant raman Raman scattering of

t-ZrO2

Inelastic X-Ray scattering

IXS of Diaspore

To model spectra need to treat scattering dynamics of incident and emergent radiation.

In case of INS interaction is between point neutron and nucleus - scalar quantity b depends only on nucleus – specific properties.

$$\frac{d^2\sigma}{dEd\Omega} = \frac{k_f}{k_i} b^2 S(\boldsymbol{Q}, \omega)$$

Q is scattering vector and ω is frequency - interact with phonons at same wavevector and frequency.

Full measured spectrum includes overtones and combinations and instrumental geometry and BZ sampling factors.

Need specific spectral modelling software to incorporate effects as postprocessing step following CASTEP phonon DOS calculation.

A-Climax : A. J. Ramirez-Cuesta Comput. Phys. Comm. 157 226 (2004))



INS of Ammonium Fluoride

LO/TO Splitting

Modelling of spectra Inelastic Neutron

Scattering INS of Ammonium

Fluoride

Single-crystal infrared Powder infrared Modelling of powder IR Non-resonant raman Raman scattering of

t-ZrO2

Inelastic X-Ray

scattering

IXS of Diaspore

- NH₄F is one of a series of ammonium halides studied in the TOSCA spectrometer. Collab. Mark Adams (ISIS)
- Structurally isomorphic with ice ih
- INS spectrum modelled using A-CLIMAX software (A. J. Ramirez Cuesta, ISIS)
- Predicted INS spectrum in mostly excellent agreement with experiment
- NH₄ libration modes in error by \approx 5%.
- Complete mode assignment achieved.







Single-crystal infrared

LO/TO Splitting

Modelling of spectra Inelastic Neutron Scattering INS of Ammonium Fluoride

Single-crystal infrared

Powder infrared Modelling of powder IR Non-resonant raman Raman scattering of t-ZrO2 Inelastic X-Ray scattering IXS of Diaspore Prediction of reflectivity of optically flat single crystal surface given as function of q- projected permittivity $\epsilon_q(\omega)$

$$R(\omega) = \left| \frac{\epsilon_q^{1/2}(\omega) - 1}{\epsilon_q^{1/2}(\omega) + 1} \right|^2$$

with ϵ_q defined in terms of ϵ^{∞} and mode oscillator strength $S_{m,\alpha\beta}$

$$\epsilon_{\boldsymbol{q}}(\omega) = \boldsymbol{q}.\boldsymbol{\epsilon}^{\boldsymbol{\infty}}.\boldsymbol{q} + \frac{4\pi}{\Omega_0} \sum_{m} \frac{\boldsymbol{q}.\boldsymbol{S}.\boldsymbol{q}}{\omega_m^2 - \omega^2} = \boldsymbol{q}.\boldsymbol{\epsilon}(\omega).\boldsymbol{q}$$





FIG. 1. (Color online) Calculated IR reflectivity spectra (blue) of a BiFeO₃ monocrystal (A), (B) and ceramics (C). Experimental data (20 K) from Ref. 12 (red).



Powder infrared

LO/TO Splitting





- Straightforward to compute peak *areas*.
- Peak shape modelling depends on sample and experimental variables.
- Multiphonon and overtone terms less straightforward.



Modelling of powder IR

LO/TO Splitting

Modelling of spectra Inelastic Neutron Scattering INS of Ammonium Fluoride Single-crystal infrared Powder infrared Modelling of powder IR

Non-resonant raman Raman scattering of t-ZrO2 Inelastic X-Ray scattering IXS of Diaspore

See E. Balan, A. M. Saitta, F. Mauri, G. Calas. First-principles modeling of the infrared spectrum of kaolinite. American Mineralogist, 2001, 86, 1321-1330. Spectral shape determined by optical effects and shifted LO modes in specific size/shape of crystallites.



FIGURE 1. Schematic view of a dielectric plate showing the electric field and polarization vectors used to calculate the IR absorption spectrum. The surface charges $+\sigma$ and $-\sigma$ produce the depolarization field, $E_{m}-E_{KB}$



FIGURE 5. Experimental (top) and theoretical (bottom) IR absorption spectra of kaolinite: (a) mid-IR range, (b) OH-stretching region. Positions of the major features are indicated by their wavenumbers. The vertical bars at the bottom correspond to the theoretical TO phonon frequencies (see Table 4), i.e., to the resonances of the imaginary part of the dielectric function. Note the shift between the absorption bands (dotted vertical bars) and the corresponding TO frequencies observed for the Si-O and OH stretching modes polarized along c.



Non-resonant raman

LO/TO Splitting

Modelling of spectra Inelastic Neutron Scattering INS of Ammonium Fluoride Single-crystal infrared Powder infrared Modelling of powder IR

Non-resonant raman Raman scattering of t-ZrO2 Inelastic X-Ray scattering IXS of Diaspore Raman scattering depends on raman activity tensor

$$I_{\alpha\beta}^{\rm raman} = \frac{d^3 E}{d\varepsilon_{\alpha} d\varepsilon_{\beta} dQ_m} = \frac{d\epsilon_{\alpha\beta}}{dQ_m}$$

i.e. the activity of a mode is the derivative of the dielectric permittivity with respect to the displacement along the mode eigenvector.

CASTEP evaluates the raman tensors using hybrid DFPT/finite displacement approach.

Raman calculation is fairly expensive \Rightarrow and is not activated by default (though group theory prediction of active modes is still performed)

Parameter calculate_raman = true in a task=phonon calculation.

Spectral modelling of IR spectrum is relatively simple function of activity.

$$\frac{d\sigma}{d\Omega} = \frac{(2\pi\nu)^4}{c^4} |\boldsymbol{e}_S.I.\boldsymbol{e}_L|^2 \frac{h(n_m+1)}{4\pi\omega_m}$$

with the thermal population factor

$$n_m = \left[exp\left(\frac{\hbar\omega_m}{k_B T}\right) - 1 \right]^{-1}$$

which is implemented in dos.pl using the -raman flag.



Raman scattering of t-ZrO2

LO/TO Splitting







Inelastic X-Ray scattering

Ana brzer





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LO/TO Splitting

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